

**Fifth International BEYOND 2010 Conference on
Beyond the Standard Models of Particle Physics,
Cosmology and Astrophysics**

Cape Town, South Africa

1 - 6, February 2010

Synthesis and Study of Superheavy Elements

Andrey Popeko

*Flerov Laboratory of Nuclear Reactions,
Joint Institute for Nuclear Research,
Dubna, Russia*

Bottom lines:

In reactions $^{48}\text{Ca} + ^{238}\text{U}$, ^{237}Np , $^{242,244}\text{Pu}$, ^{243}Am , $^{245,248}\text{Cm}$, ^{249}Bk and ^{249}Cf :

- Isotopes $^{282-286}113$, $^{286-289}114$, $^{287-290}115$, $^{290-293}116$, $^{293,294}117$ and $^{294}118$ of 6 superheavy elements have been synthesized,
- Heavy isotopes of known elements ^{267}Rf , $^{266-268,270}\text{Db}$, $^{266,271}\text{Sg}$, $^{270-272,274}\text{Bh}$, $^{270,275}\text{Hs}$, $^{274-276,278}\text{Mt}$, $^{279,281}\text{Ds}$, $^{278-282}\text{Rg}$ and $^{282-285}\text{Cn}$ have been synthesized,
- First chemical experiments with elements 112 and 114 have been conducted.

Flerov Laboratory of Nuclear Reactions



BASIC DIRECTIONS of RESEARCH

1. Heavy and superheavy nuclei

- **Synthesis and study of properties of superheavy elements**
- **Chemistry of new elements**
- **Fusion-fission and multi-nucleon transfer reactions**
- **Mass-spectrometry and nuclear spectroscopy of SH nuclei**

2. Light exotic nuclei

- **Properties and structure of light exotic nuclei**
- **Reactions with exotic nuclei**

3. Radiation effects and physical bases of nanotechnology

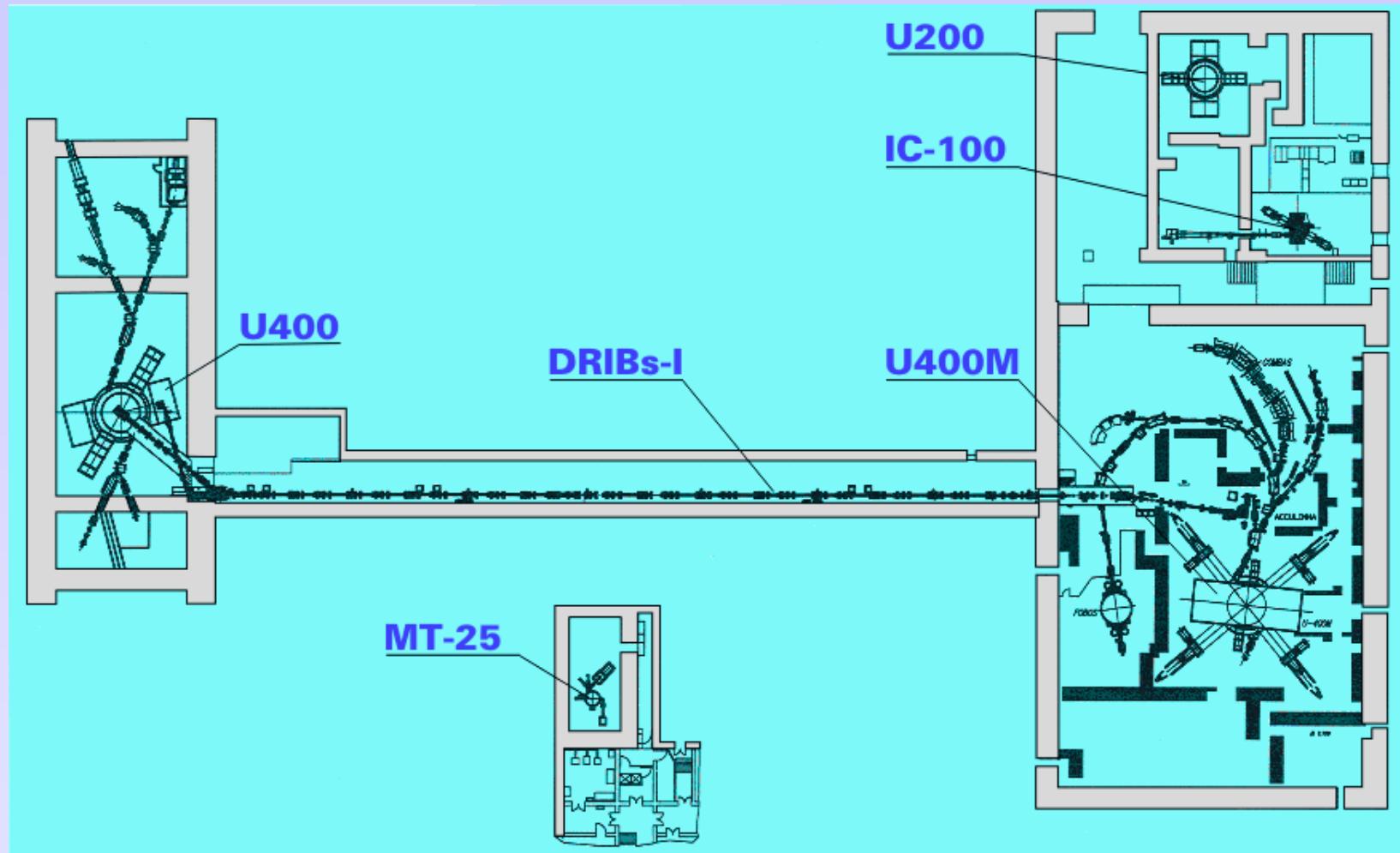
4. Accelerator technology

Staff : 350 people (including 100 younger than 35 years old)

Budget: ~ 15.0 M\$

Out of budget staff: 140 people

FLNR accelerators



Periodic table of the elements

Dmitri Mendeleev (1869)

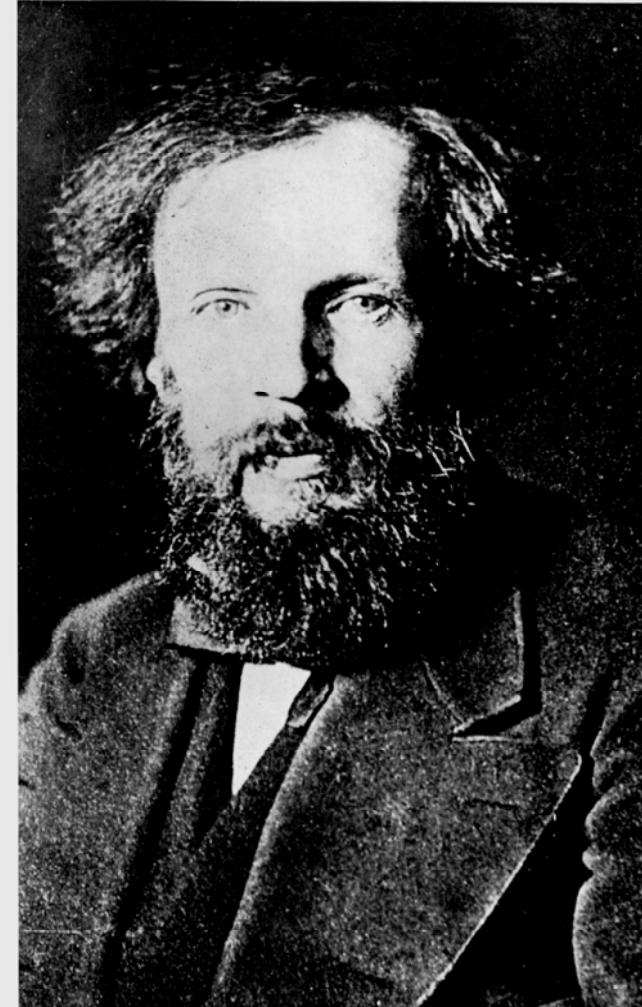
Изобретение периодической системы химических элементов с единообразными свойствами
Д. Менделеевым

$Z = 8$	$Z = 10$	$Z = 18$
$Z = 11$	$Z = 12$	$Z = 16$
$Z = 14$	$Z = 15$	$Z = 17$
$Z = 16$	$Z = 17$	$Z = 18$
$Z = 19$	$Z = 20$	$Z = 22$
$Z = 22$	$Z = 23$	$Z = 24$
$Z = 24$	$Z = 25$	$Z = 26$
$Z = 26$	$Z = 27$	$Z = 28$
$Z = 28$	$Z = 29$	$Z = 30$
$Z = 30$	$Z = 31$	$Z = 32$
$Z = 31$	$Z = 32$	$Z = 33$
$Z = 32$	$Z = 33$	$Z = 34$
$Z = 33$	$Z = 34$	$Z = 36$
$Z = 34$	$Z = 35$	$Z = 37$
$Z = 35$	$Z = 36$	$Z = 38$
$Z = 36$	$Z = 37$	$Z = 39$
$Z = 37$	$Z = 38$	$Z = 40$
$Z = 38$	$Z = 39$	$Z = 41$
$Z = 39$	$Z = 40$	$Z = 42$
$Z = 40$	$Z = 41$	$Z = 43$
$Z = 41$	$Z = 42$	$Z = 44$
$Z = 42$	$Z = 43$	$Z = 45$
$Z = 43$	$Z = 44$	$Z = 46$
$Z = 44$	$Z = 45$	$Z = 47$
$Z = 45$	$Z = 46$	$Z = 48$
$Z = 46$	$Z = 47$	$Z = 49$
$Z = 47$	$Z = 48$	$Z = 50$
$Z = 48$	$Z = 49$	$Z = 51$
$Z = 49$	$Z = 50$	$Z = 52$
$Z = 50$	$Z = 51$	$Z = 53$
$Z = 51$	$Z = 52$	$Z = 54$
$Z = 52$	$Z = 53$	$Z = 55$
$Z = 53$	$Z = 54$	$Z = 56$
$Z = 54$	$Z = 55$	$Z = 57$
$Z = 55$	$Z = 56$	$Z = 58$
$Z = 56$	$Z = 57$	$Z = 59$
$Z = 57$	$Z = 58$	$Z = 60$
$Z = 58$	$Z = 59$	$Z = 61$
$Z = 59$	$Z = 60$	$Z = 62$
$Z = 60$	$Z = 61$	$Z = 63$
$Z = 61$	$Z = 62$	$Z = 64$
$Z = 62$	$Z = 63$	$Z = 65$
$Z = 63$	$Z = 64$	$Z = 66$
$Z = 64$	$Z = 65$	$Z = 67$
$Z = 65$	$Z = 66$	$Z = 68$
$Z = 66$	$Z = 67$	$Z = 69$
$Z = 67$	$Z = 68$	$Z = 70$
$Z = 68$	$Z = 69$	$Z = 71$
$Z = 69$	$Z = 70$	$Z = 72$
$Z = 70$	$Z = 71$	$Z = 73$
$Z = 71$	$Z = 72$	$Z = 74$
$Z = 72$	$Z = 73$	$Z = 75$
$Z = 73$	$Z = 74$	$Z = 76$
$Z = 74$	$Z = 75$	$Z = 77$
$Z = 75$	$Z = 76$	$Z = 78$
$Z = 76$	$Z = 77$	$Z = 79$
$Z = 77$	$Z = 78$	$Z = 80$
$Z = 78$	$Z = 79$	$Z = 81$
$Z = 79$	$Z = 80$	$Z = 82$
$Z = 80$	$Z = 81$	$Z = 83$
$Z = 81$	$Z = 82$	$Z = 84$
$Z = 82$	$Z = 83$	$Z = 85$
$Z = 83$	$Z = 84$	$Z = 86$
$Z = 84$	$Z = 85$	$Z = 87$
$Z = 85$	$Z = 86$	$Z = 88$
$Z = 86$	$Z = 87$	$Z = 89$
$Z = 87$	$Z = 88$	$Z = 90$
$Z = 88$	$Z = 89$	$Z = 91$
$Z = 89$	$Z = 90$	$Z = 92$
$Z = 90$	$Z = 91$	$Z = 93$
$Z = 91$	$Z = 92$	$Z = 94$
$Z = 92$	$Z = 93$	$Z = 95$
$Z = 93$	$Z = 94$	$Z = 96$
$Z = 94$	$Z = 95$	$Z = 97$
$Z = 95$	$Z = 96$	$Z = 98$
$Z = 96$	$Z = 97$	$Z = 99$
$Z = 97$	$Z = 98$	$Z = 100$

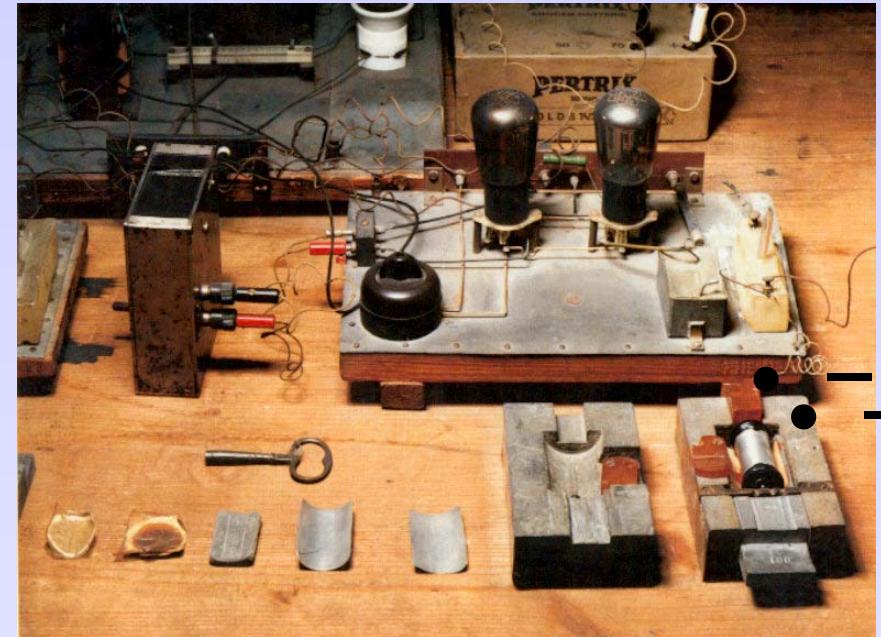
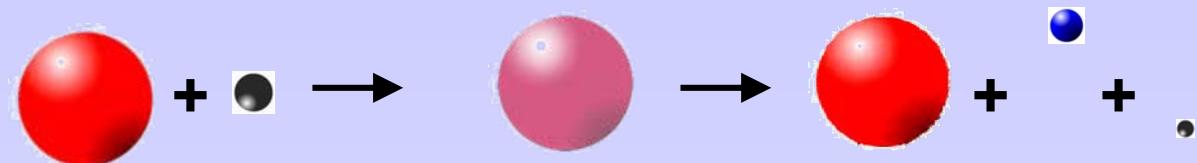
Essai d'une système des éléments d'après leurs poids atomiques et fonctions chimiques par D. Mendeleev
professeur de chimie à l'université de Saint-Pétersbourg

18 II 69.

Академик профессор Д. Менделеев

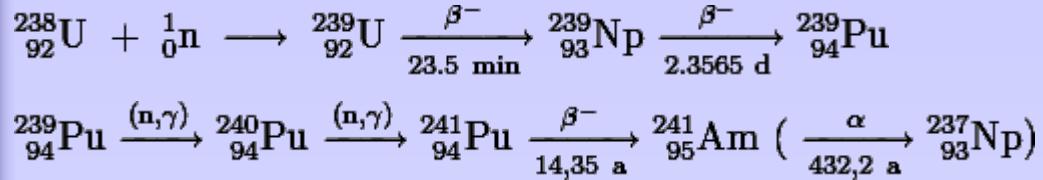


1934: search for transurane

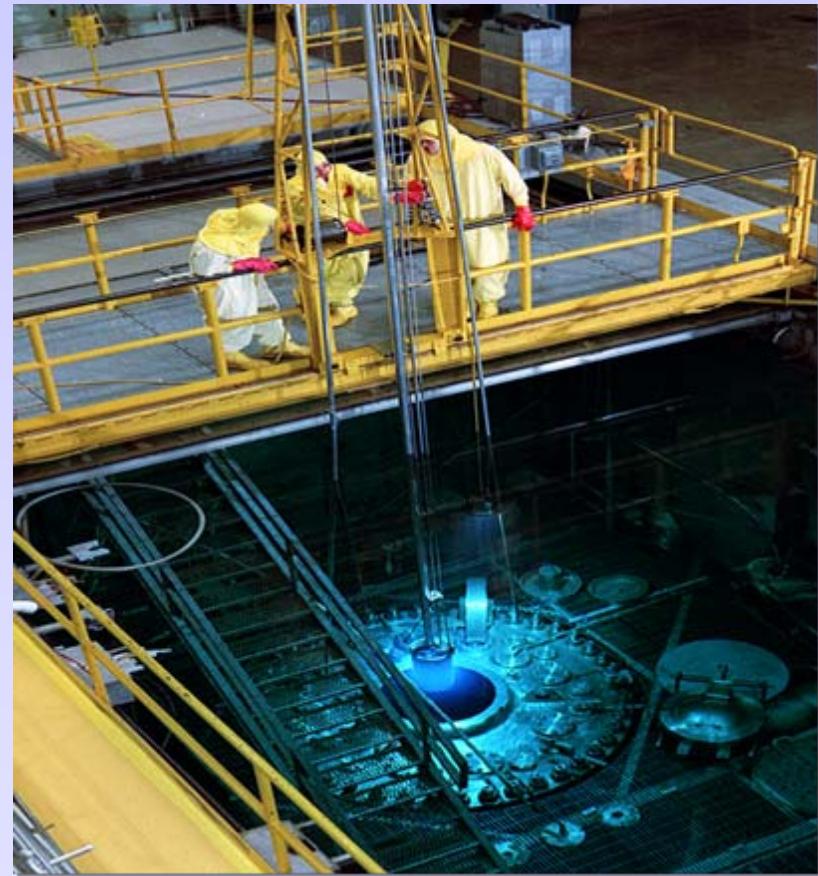


Otto Hahn und Lise Meitner

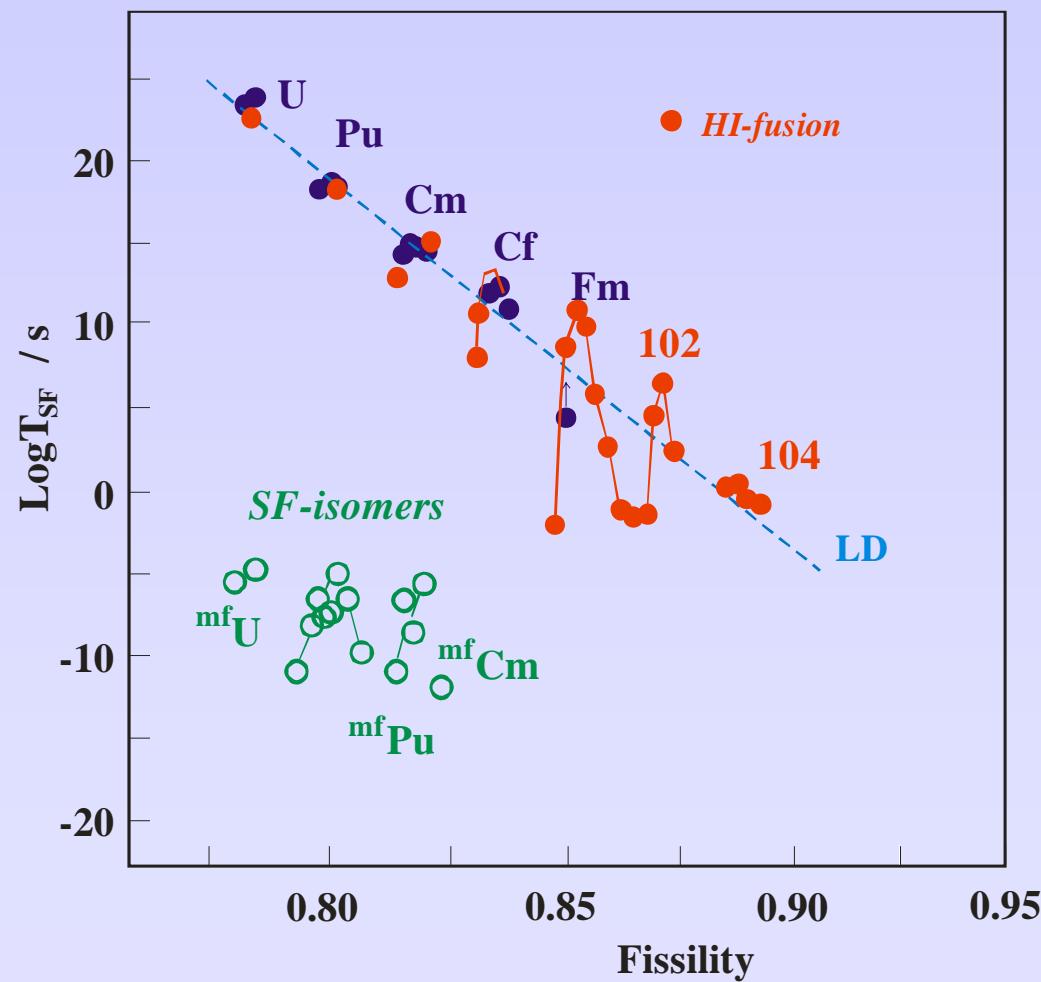
Irradiation of targets at HFIR reactor (Oak Ridge)



- Irradiation in the HFIR flux trap
 - Thermal-neutron flux of 2.5×10^{15} neutrons/cm²·s
 - 31 target positions (10–13 targets typically irradiated)
 - Produces ~35 mg ^{252}Cf per target (smaller quantities of Bk, Es, Fm, others)



Spontaneous fission half-lives of actinides



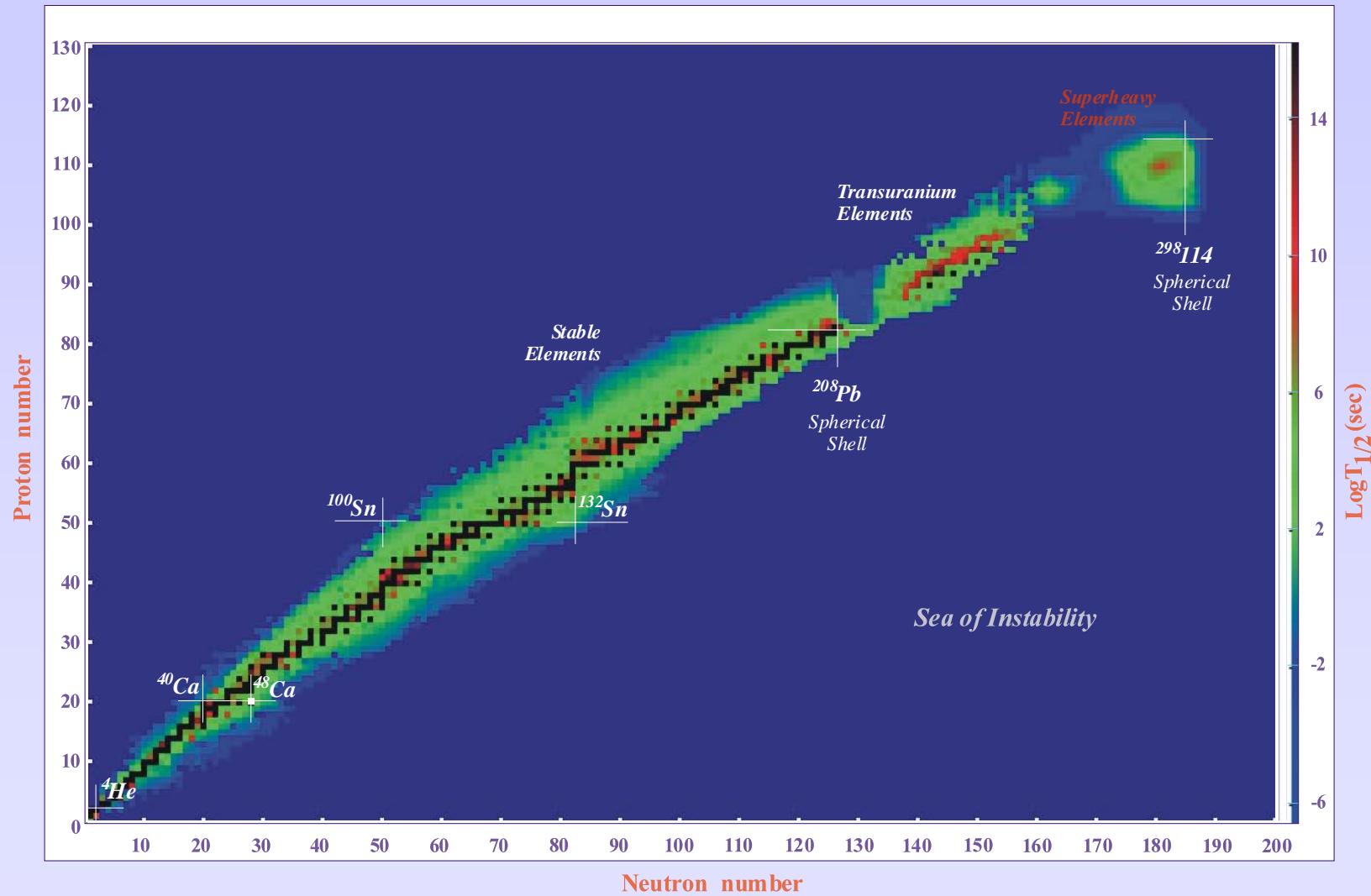
Prehistory

- 1966: A. Sobiczewski, F.A. Gareev, B.N. Kalinkin: next “magic numbers” are $Z=114$, $N=184$;
- 1966: W.D. Myers, W.J. Swiatecki: next “magic numbers” are $Z=126$, $N=184$
- 1966: V.M. Strutinsky; “shell correction” method
- 1967: H.B. Meldner: next “magic numbers” are $Z=114$, $N=184$

Accuracy of predictions:

- Spontaneous fission half-life: $T_{1/2} * 10^{\pm 10}$!!
- α -decay: $T_{1/2} * 10^{\pm 10}$!!

Chart of the Nuclides



Search for SHE in Nature

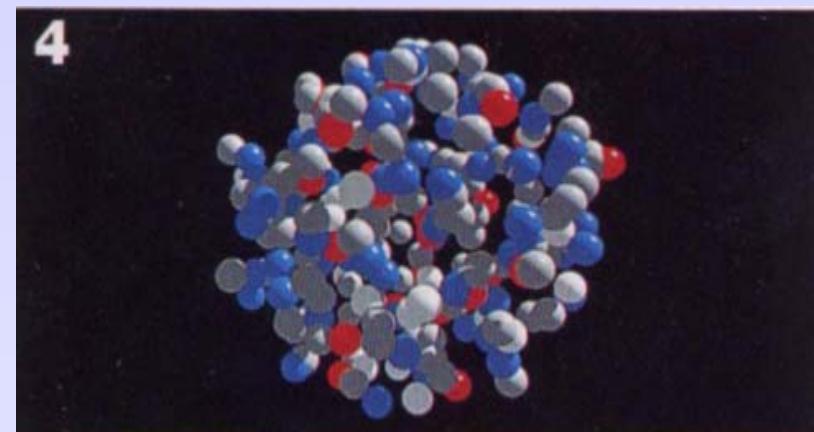
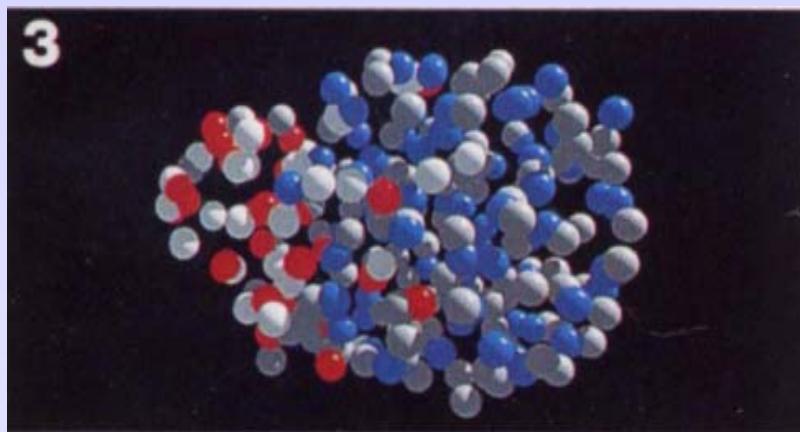
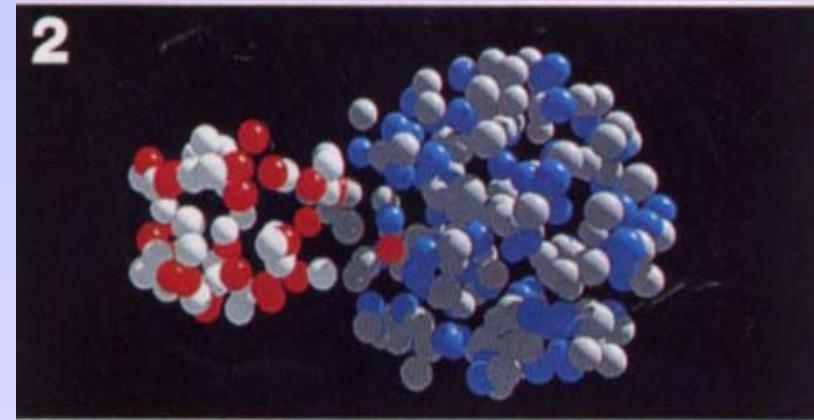
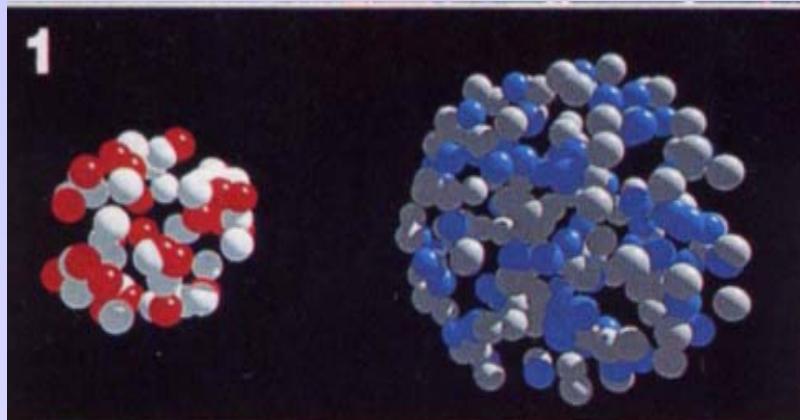
- **Search for SHE in terrestrial matter**
- **Search for SHE in meteorites**
- **Search for SHE in cosmic rays**
- **Investigation of isotopic anomalies**

20 years → void result,

**but many new high sensitive detection
methods were developed**

Complete Fusion $\text{U}+\text{N}\rightarrow\text{Es}$

instead of $\text{U}+14\text{n}\rightarrow 7\beta\rightarrow\text{Es}$



Problems: crosssection from 100 b \rightarrow 1 pb, amount from 1000 kg \rightarrow 1 mg

Synthesis of SHE at accelerators

- 1971; Orsay, France; $^{232}\text{Th} + ^{82}\text{Kr} \rightarrow ^{310}\text{126} + 4\text{n}$; $\sigma_{4\text{n}} < 0.5 \text{ mb} !!!$
- 1971-1975; Dubna, SU; deep inelastic or fission reactions of ^{76}Ge , $^{136}\text{Xe} + ^{238}\text{U}$
- 1975; Dubna, SU; $^{48}\text{Ca} + \text{actinides}$

Cross-section limits for ^{48}Ca -induced reactions

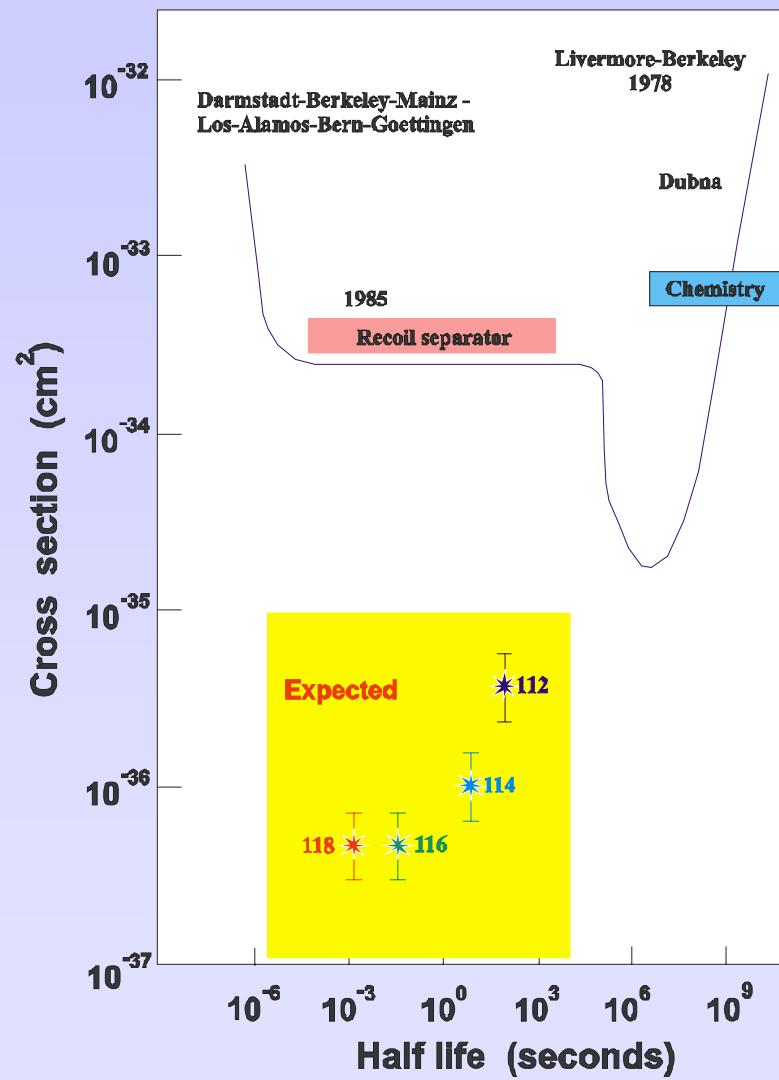
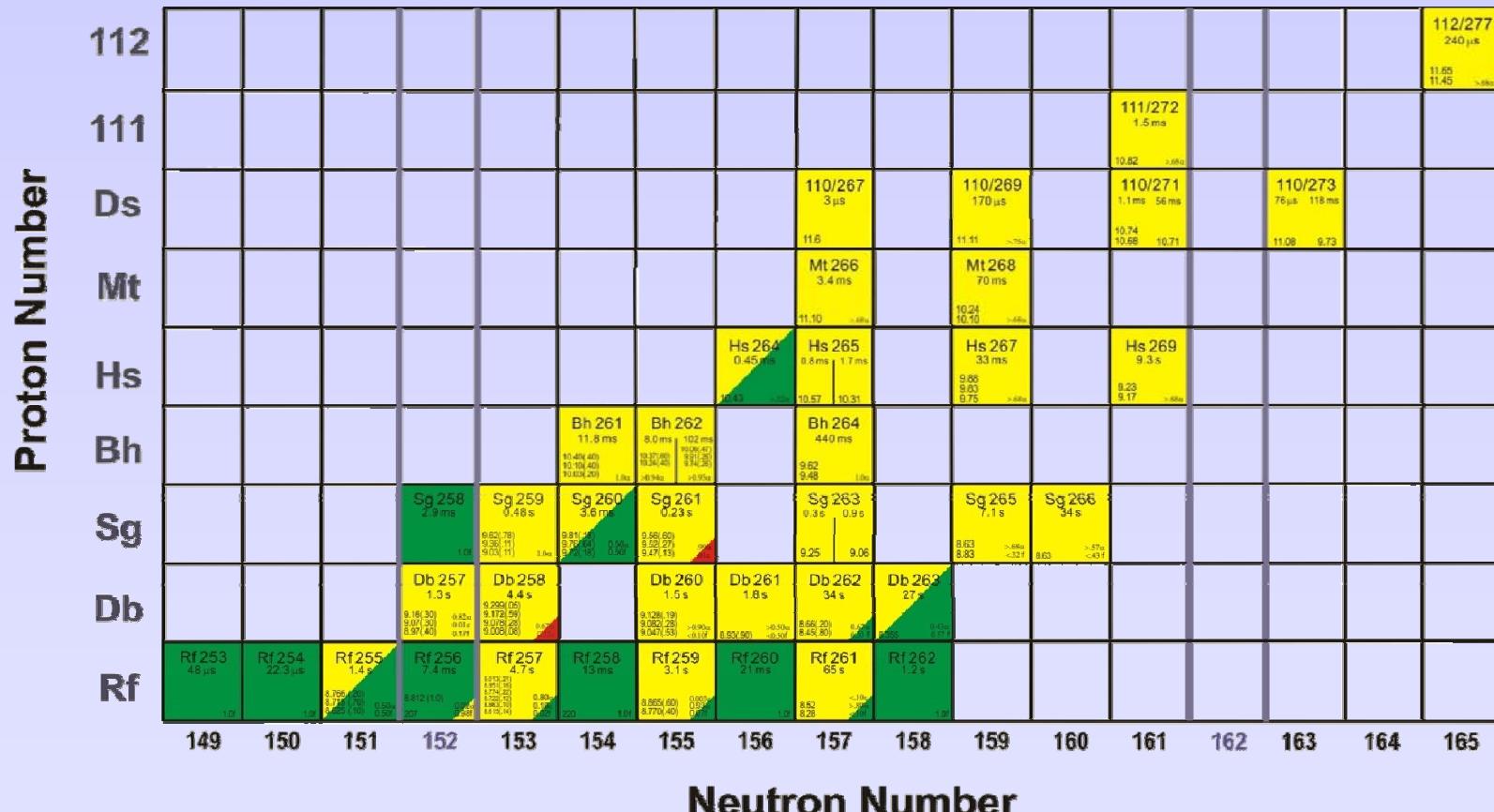
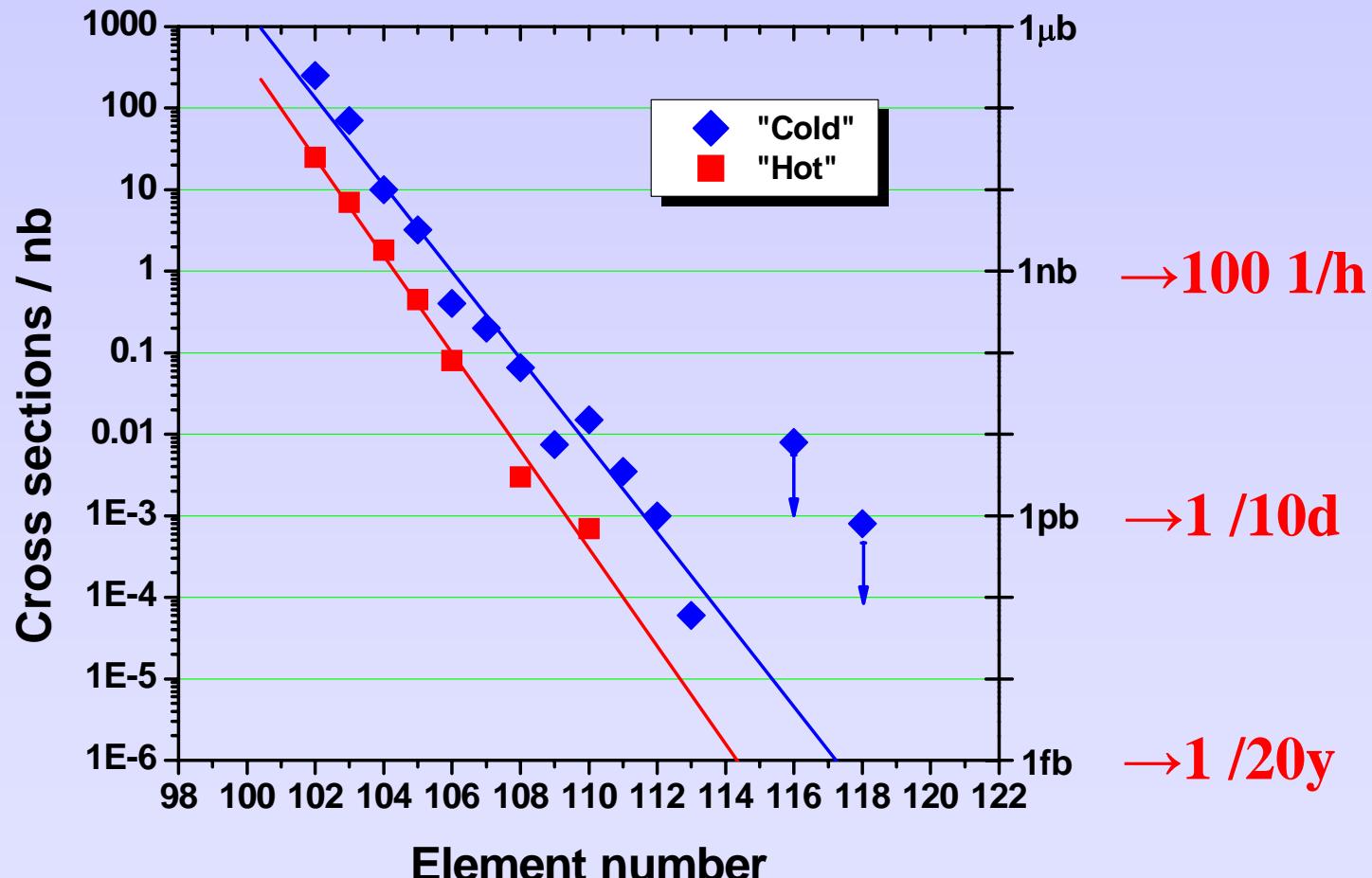


Chart of transactinide nuclides 1996



Cross-sections of “Hot” (actinide targets) and “Cold” (Pb or Bi targets) fusion reactions



“Warm” ($^{48}\text{Ca} + \text{actinide targets}$) complete fusion reactions

Efforts focused on the synthesis of SHE

New ECR-ion source
(GANIL, JINR)

^{48}Ca

ACCELERATORS

ISOTOPE
ENRICHMENT

Enrichment up to 68-70%
(Lesnoy)

beam
intensity - $4 - 8 \cdot 10^{12}/\text{s}$

Isotopes:

$\text{U}[233, 238]$, $\text{Pu}[242, 244]$, $\text{Am}[243]$, $\text{Cm}[245, 248]$, $\text{Cf}[249] + ^{48}\text{Ca} \rightarrow Z = 112 - 118$

technology of the target
preparation – 0.3 mg/cm^2

beam time - 4000 h/y

isotope enrichment 98-99%
S-2 separator
(Sarov)

Separation and detection of
superheavy nuclei

REACTOR
REGIME

isotope production
high flux reactors
(Oak Ridge, Dimitrovgrad)

Gas-filled separator & detectors
(Dubna, Livermore)

Prices per 1 mg
 $^{197}\text{Au} \approx 0.03 \text{ US\$}$
 $^{239}\text{Pu} \approx 4 \text{ US\$}$
 $^{48}\text{Ca} \approx 80 \text{ US\$}$
 $^{249}\text{Cf} \approx 60,000 \text{ US\$}$

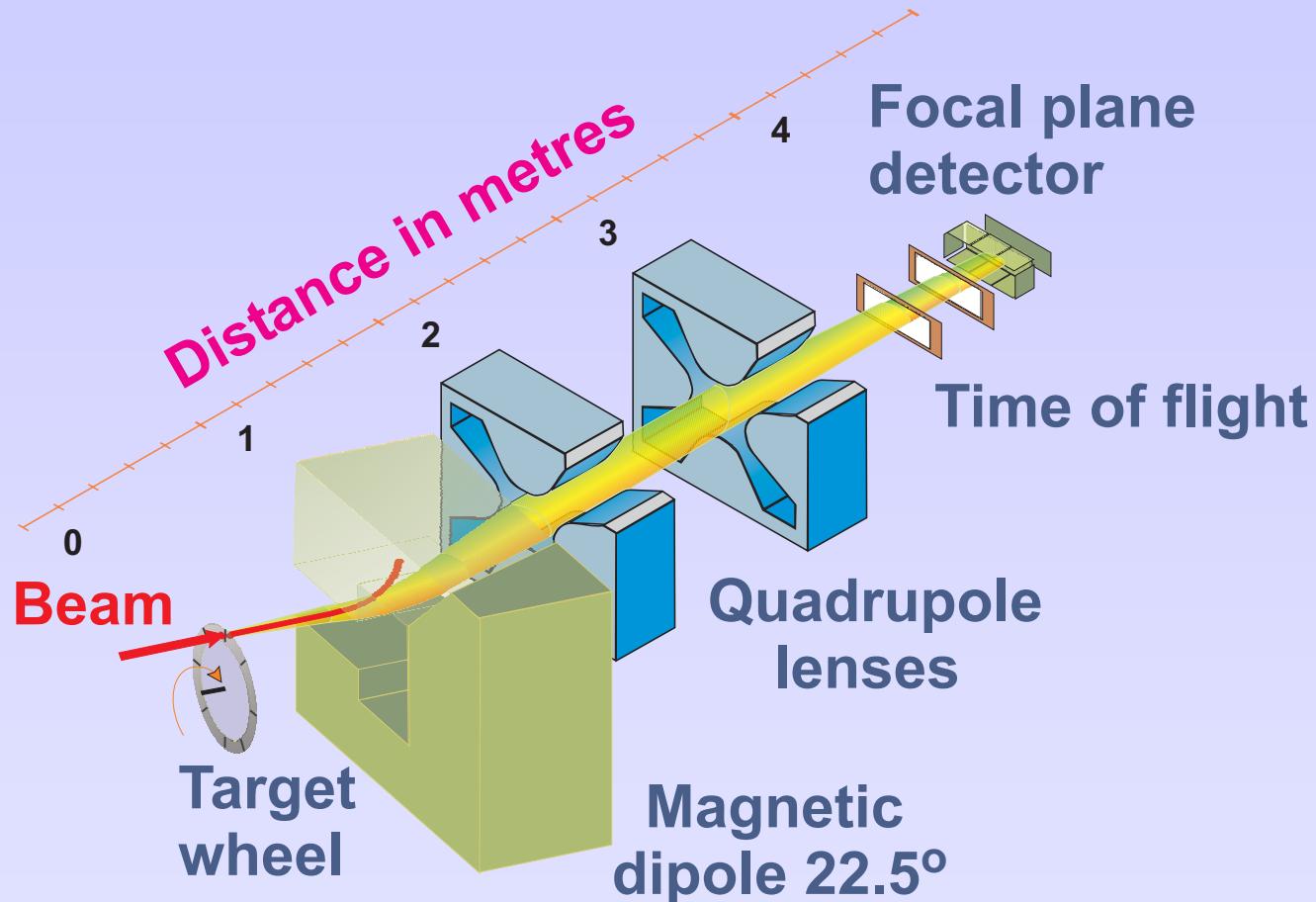
FLNR U400 cyclotron



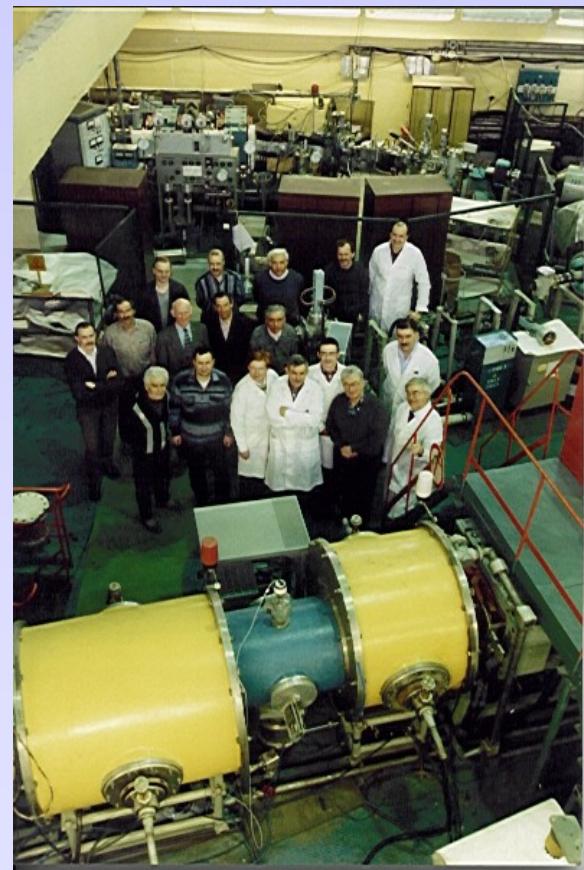
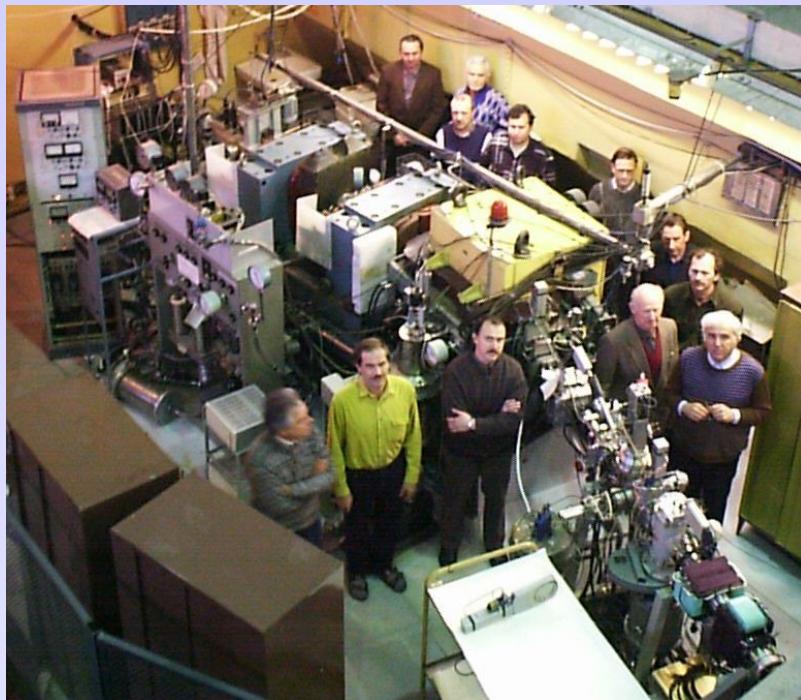
^{249}Cf - target



Dubna Gas Filled Recoil Separator

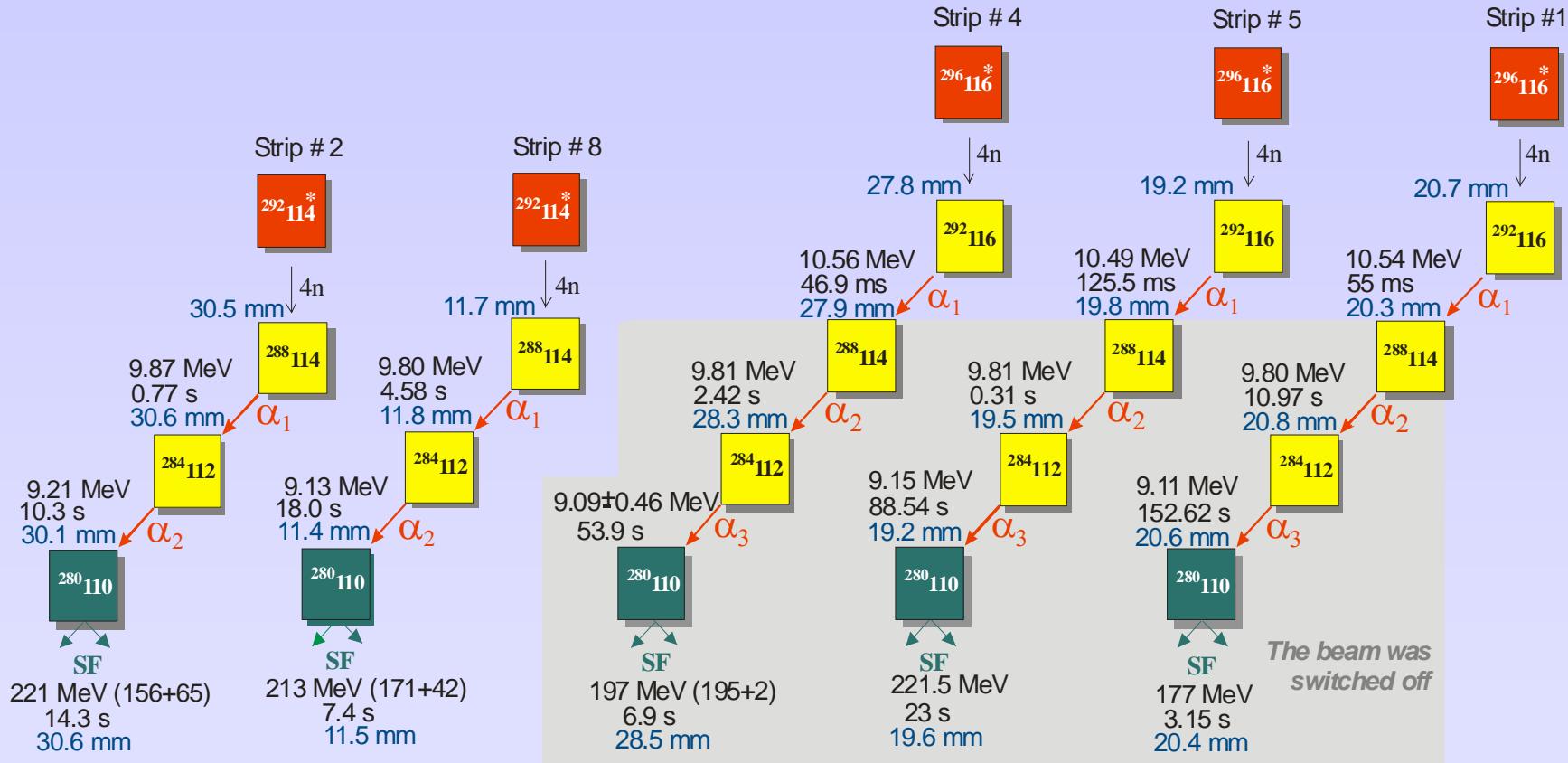


Research groups at DGFRS and VASSILISSA





Total beam dose: $1.5 \cdot 10^{19}$



June 25, 1999 05:39

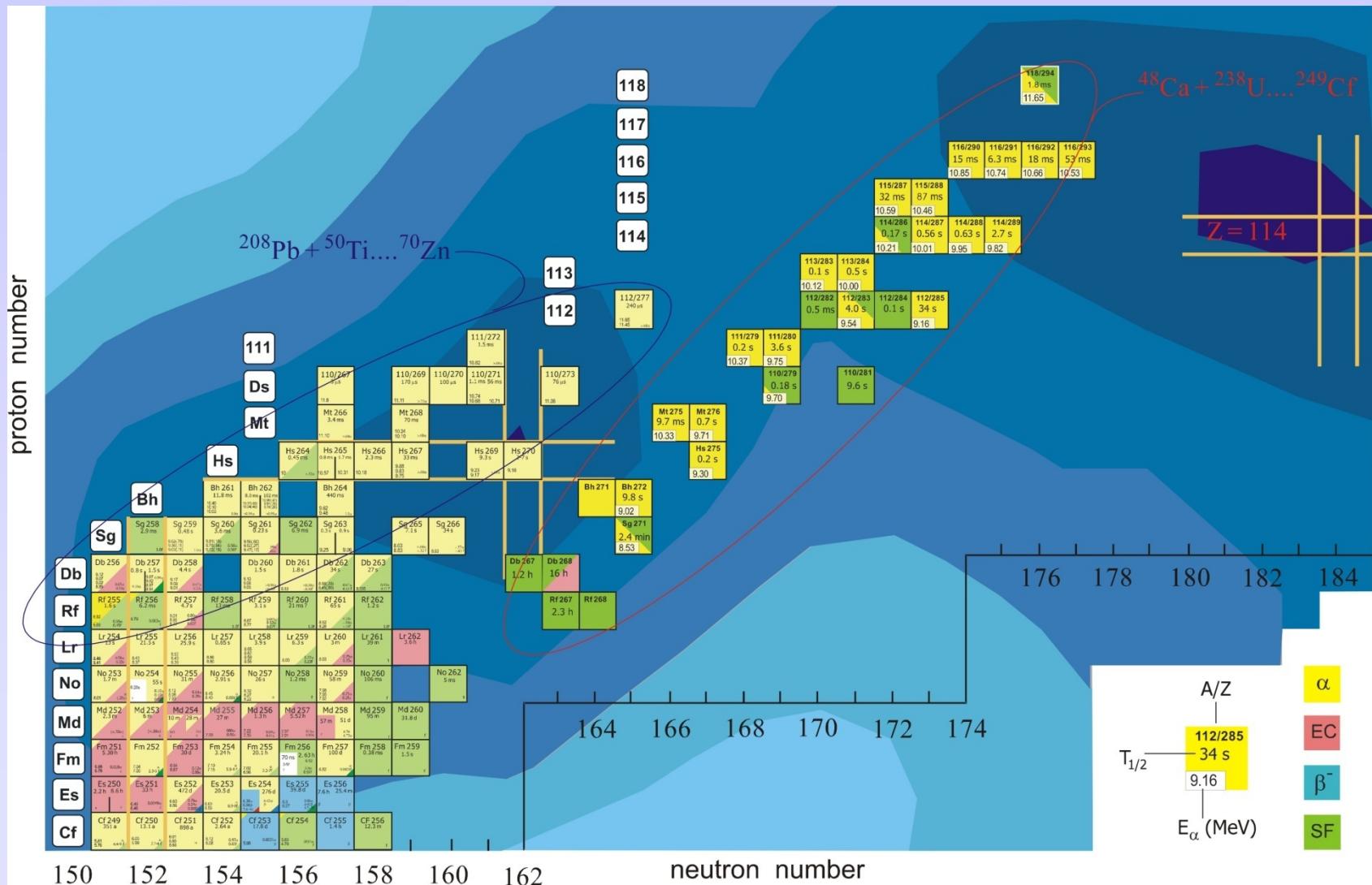
Oct. 28, 1999 22:24

July 19, 2000 01:21

May 02, 2001 06:21

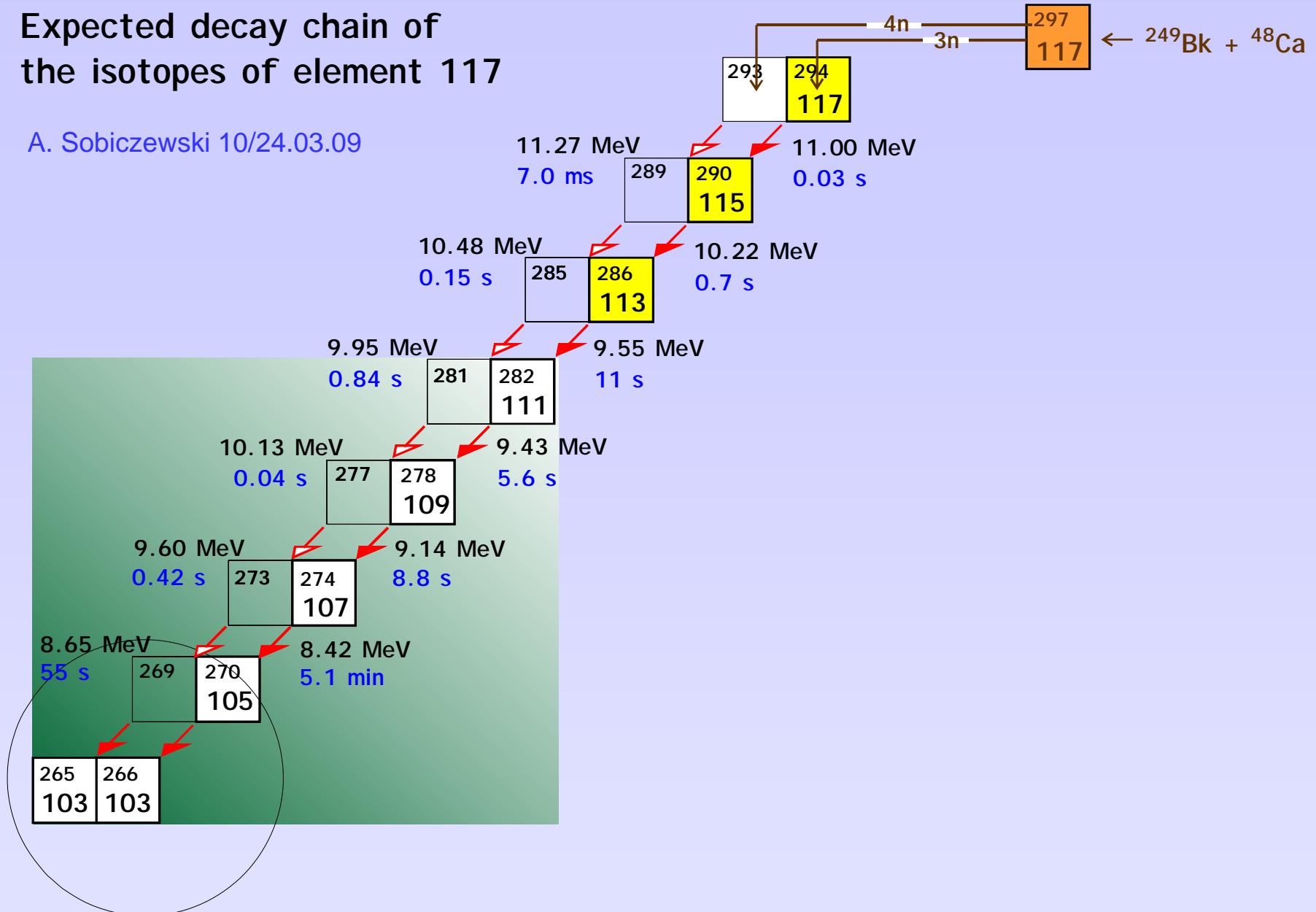
May 08, 2001 16:54

Chart of the nuclides 2006



Expected decay chain of the isotopes of element 117

A. Sobiczewski 10/24.03.09



22 mg of ^{249}Bk have been produced with neutrons of HIFR ORNL



Bk(NO₃)₃ Product

22 mg ^{249}Bk transport

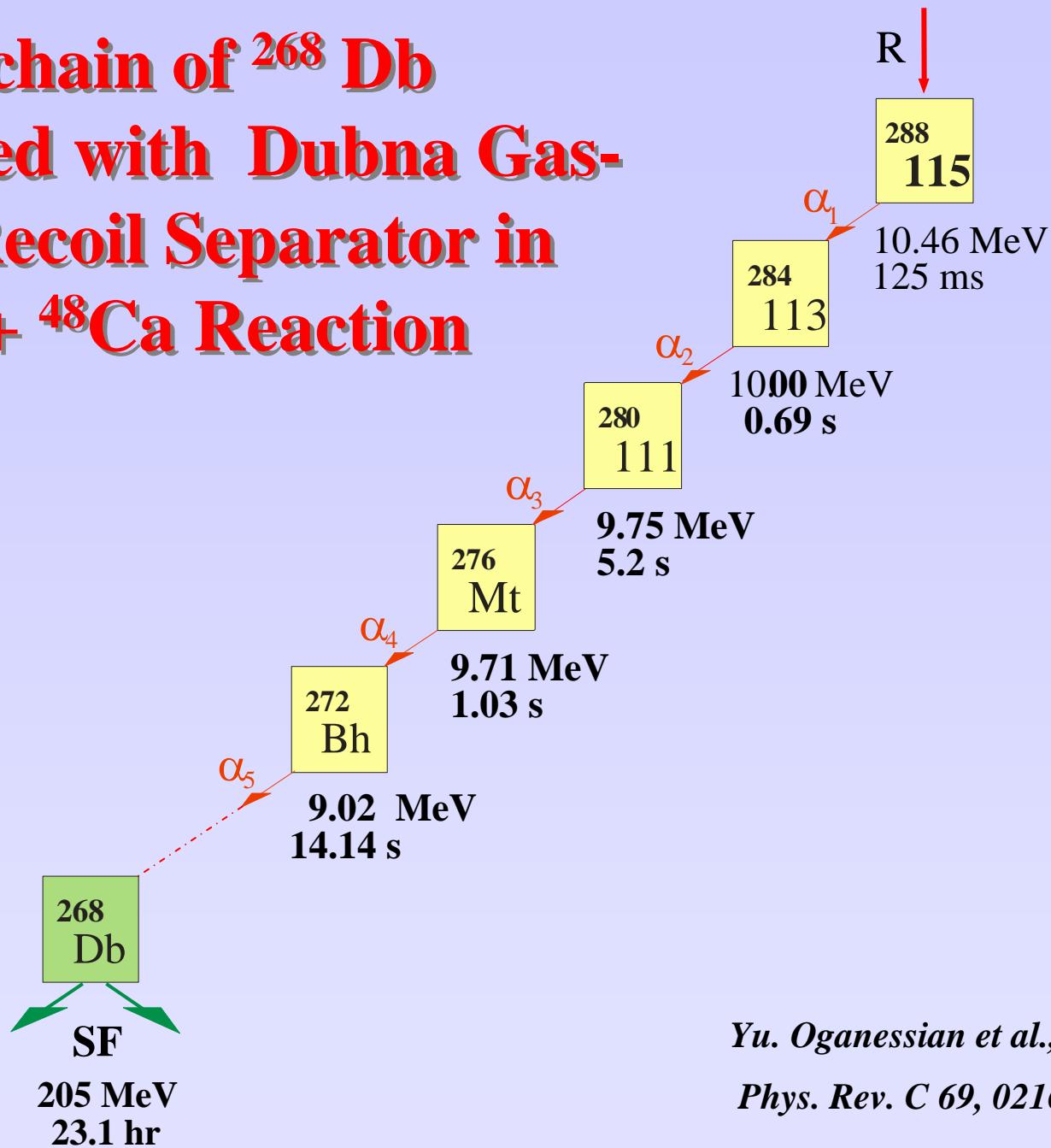


Since July 2009:

Run #	E*	∫ - flux	N _{events}
1	39 MeV	2.4×10 ¹⁹	Will be published soon
2	35 MeV	2.0×10 ¹⁹	

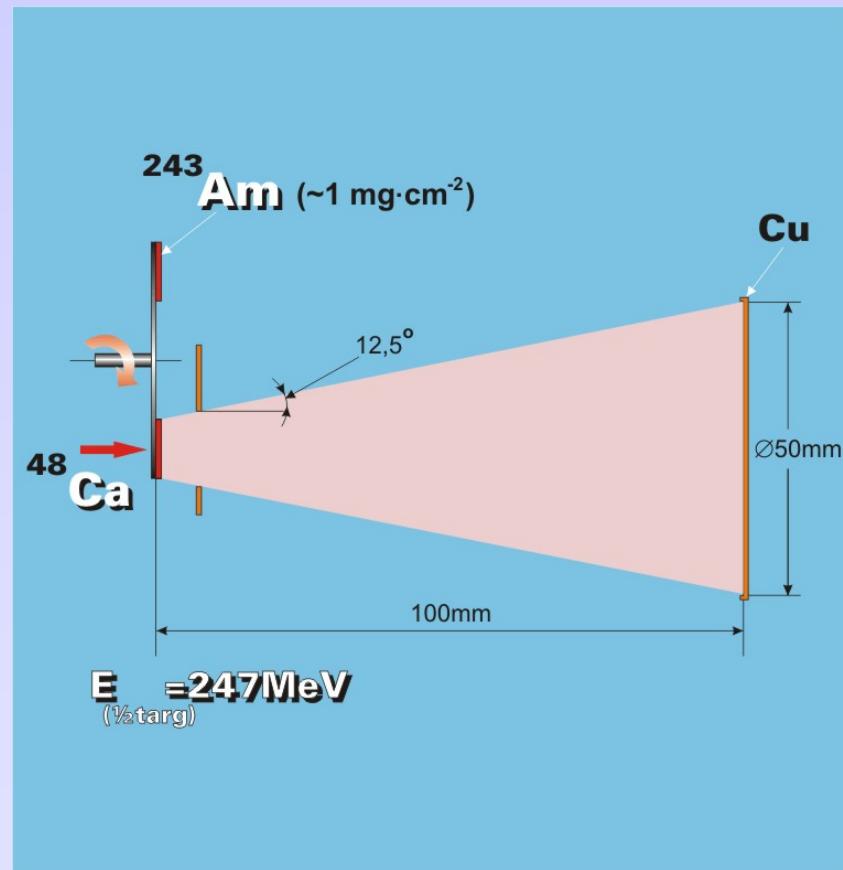
Experiment is still going on...

Decay chain of ^{268}Db observed with Dubna Gas- filled Recoil Separator in $^{243}\text{Am} + ^{48}\text{Ca}$ Reaction



Yu. Oganessian et al.,
Phys. Rev. C 69, 021601(R) 2004

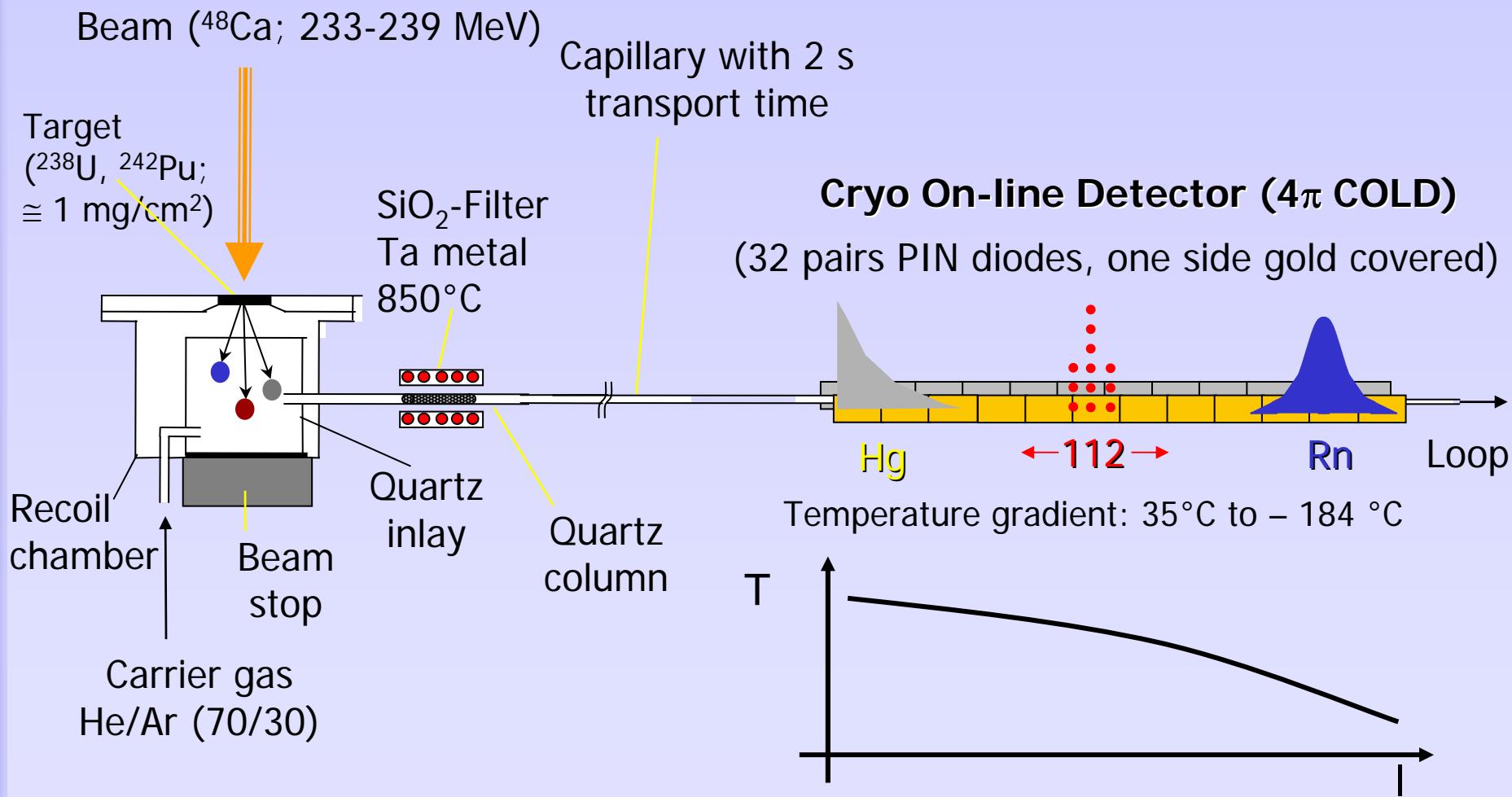
Irradiation of ^{243}Am -target with ^{48}Ca -ions



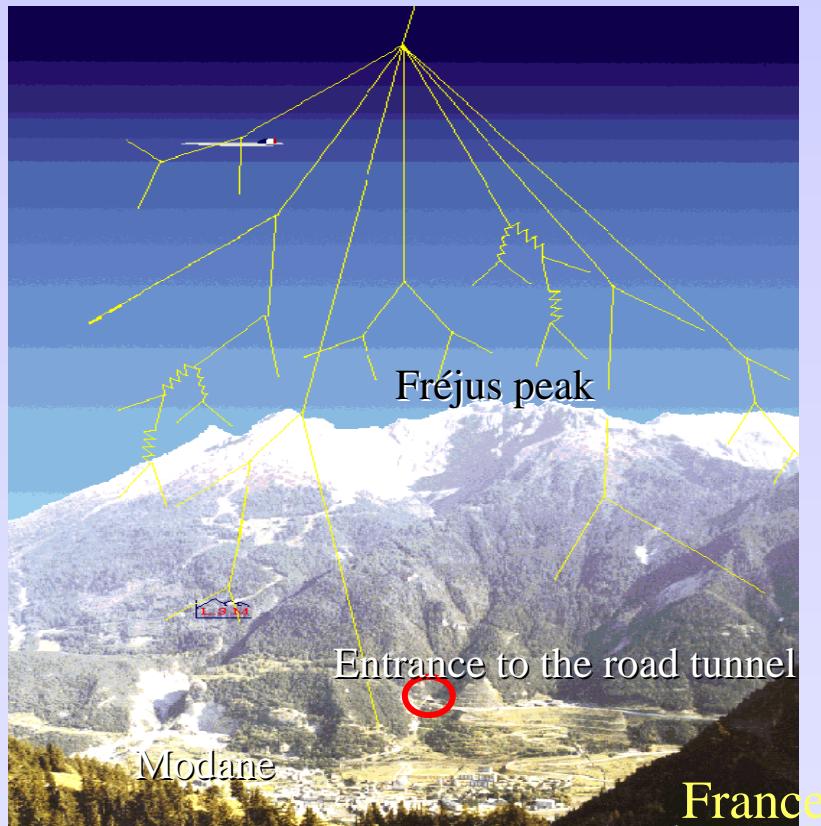
GAS PHASE CHEMISTRY WITH ELEMENTS 112 AND 114

- Are elements 112 and 114 volatile metals?
- How do relativistic effects influence the chemistry of E112 and of E114?

Chemistry of the elements 112 & 114



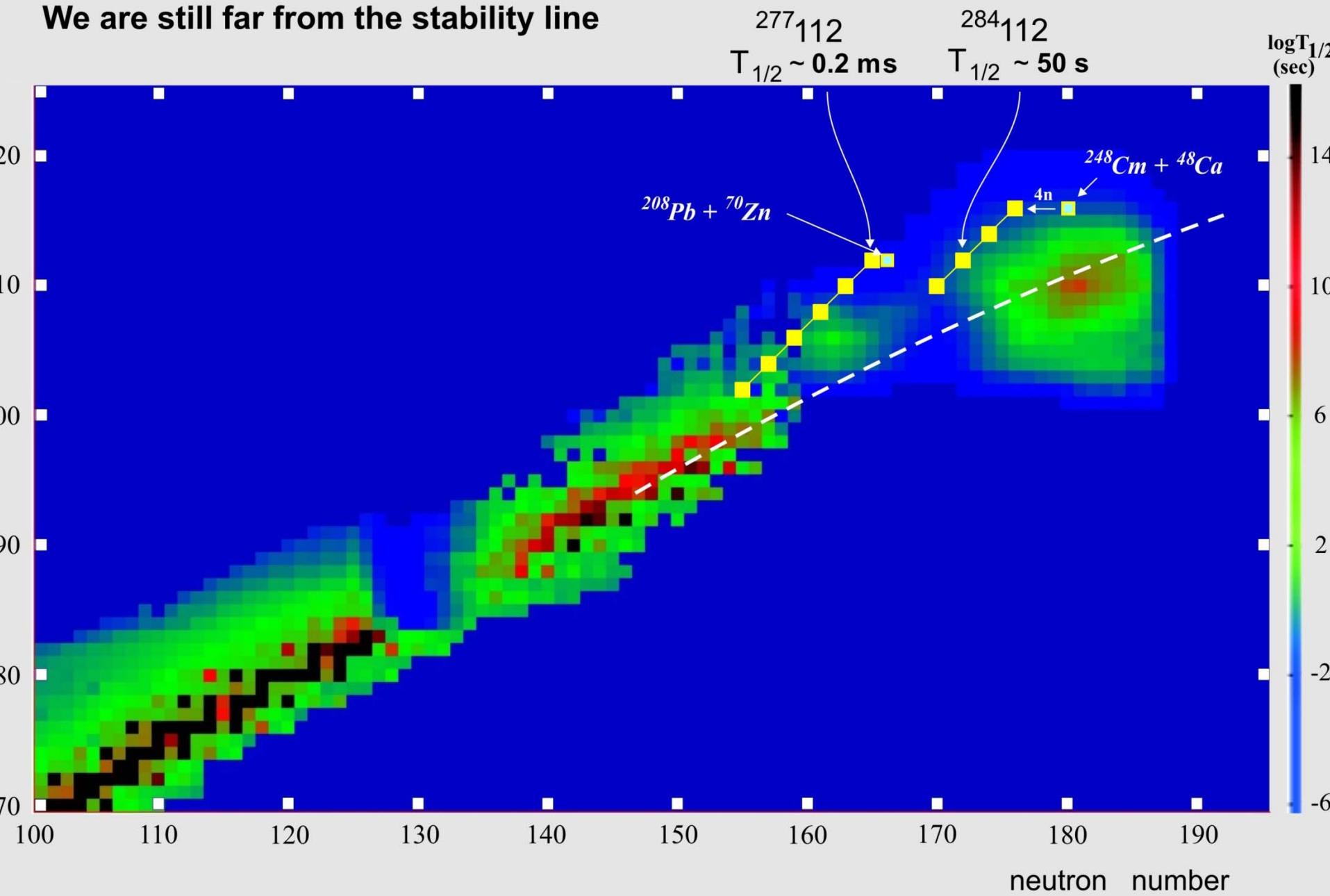
Detector installed in Modane (France)



Conclusion I

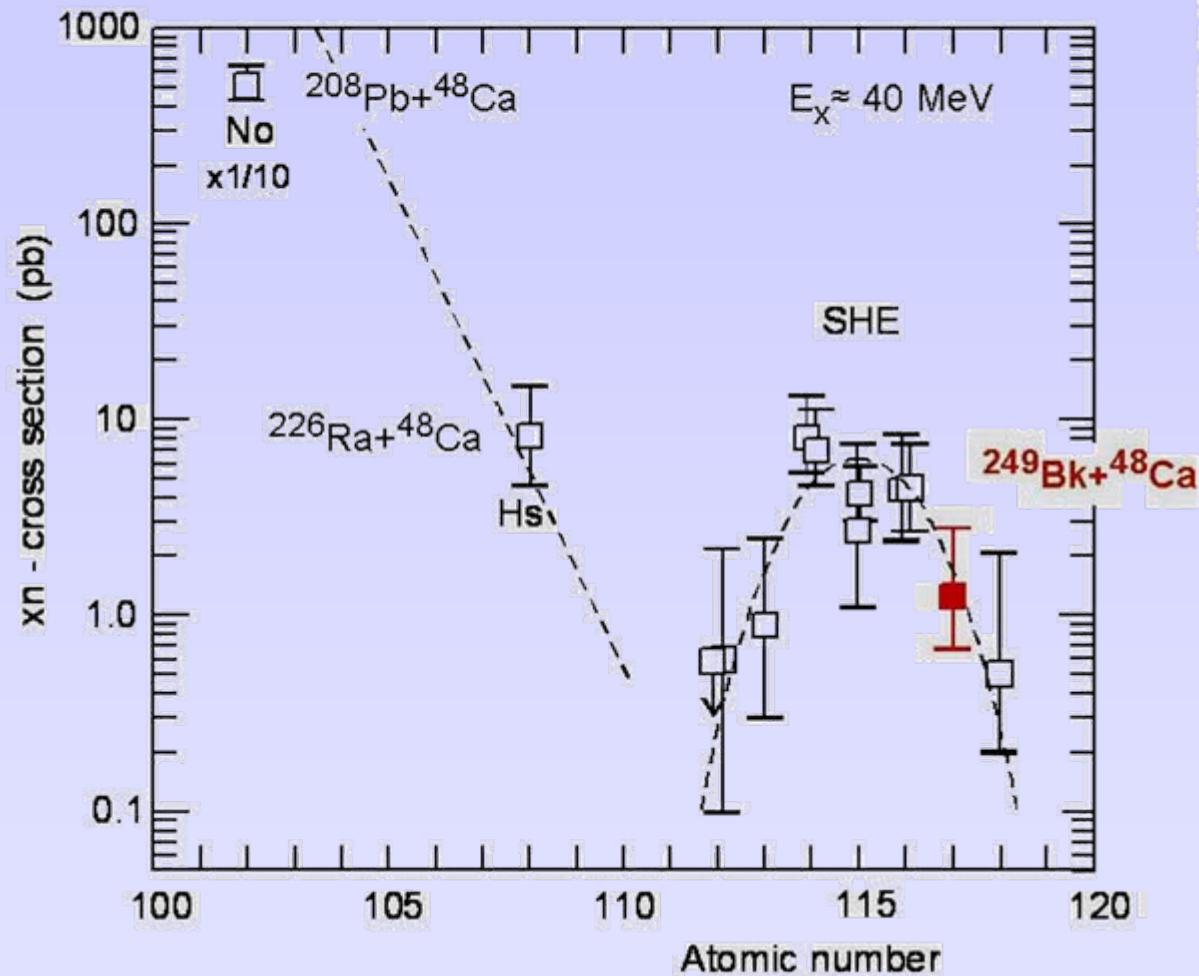
shift with 7 neutrons gives 6 orders of magnitude in half-life

We are still far from the stability line



Conclusion II

Cross sections



Yu. Oganessian 2010

Conclusion III

- Heaviest target: $^{249}\text{Cf} \rightarrow Z_{\max} = 118$;
- Projectiles: ^{54}Cr , ^{50}Ti , ^{58}Fe , ^{64}Ni ;
- symmetric reactions: $^{136}\text{Xe} + ^{150}\text{Nd}$;
- RIB: ^{132}Sn .

$^{238}\text{U} + ^{64}\text{Ni}$ GSI 2008

$^{244}\text{Pu} + ^{58}\text{Fe}$ FLNR 2007

$^{248}\text{Cm} + ^{54}\text{Cr}$ GSI 2010

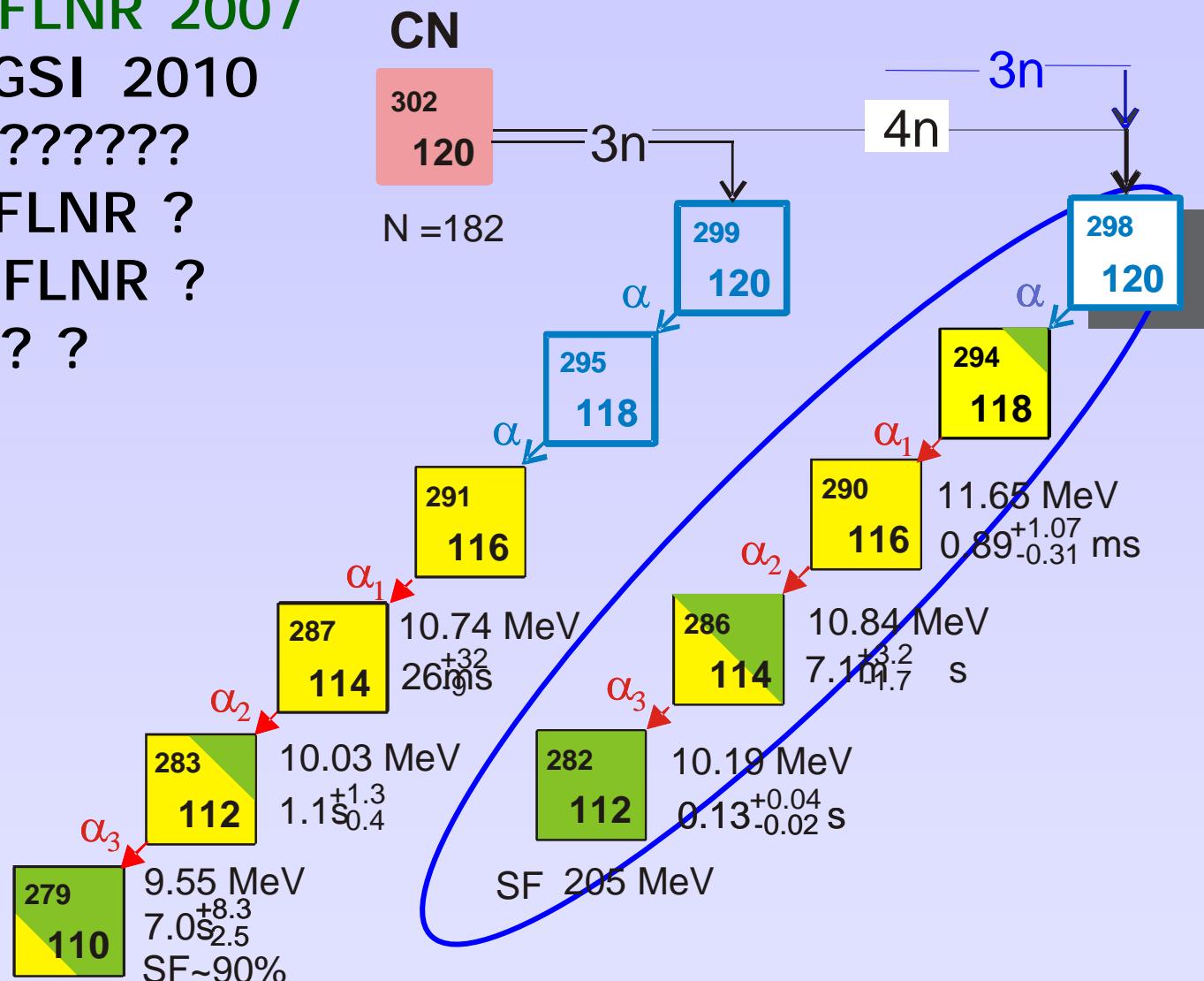
$^{252}\text{Cf} + ^{50}\text{Ti}$????????

$^{251}\text{Cf} + ^{50}\text{Ti}$ FLNR ?

$^{251}\text{Cf} + ^{48}\text{Ca}$ FLNR ?

$^{254}\text{Es} + ^{48}\text{Ca}$? ?

\downarrow
 ^{251}Cf
($\alpha/898\text{y}$)



Conclusion IV

Mendeleev periodic table of the elements (2009) (preliminary)

Mendeleyev periodic table of the elements (2009)																	
	IA																VIIIA
Boron	1	H															
	1.38794	Hydrogen															

Лантаноиды Lanthanides

Ce	58	Pr	59	Nd	60	Pm	61	Sm	62	Eu	63	Gd	64	Tb	65	Dy	66	Ho	67	Er	68	Tm	69	Yb	70	Lu	71
140,115 Dysprosium		140,97695 Neodymium		144,24 Promethium		148 Protactinium		150,38 Samarium		151,905 Europium		157,25 Gadolinium		158,0234 Terbium		162,30 Dysprosium		164,02025 Holmium		167,26 Erbium		168,02401 Thulium		173,94 Ytterbium		174,907 Lutetium	

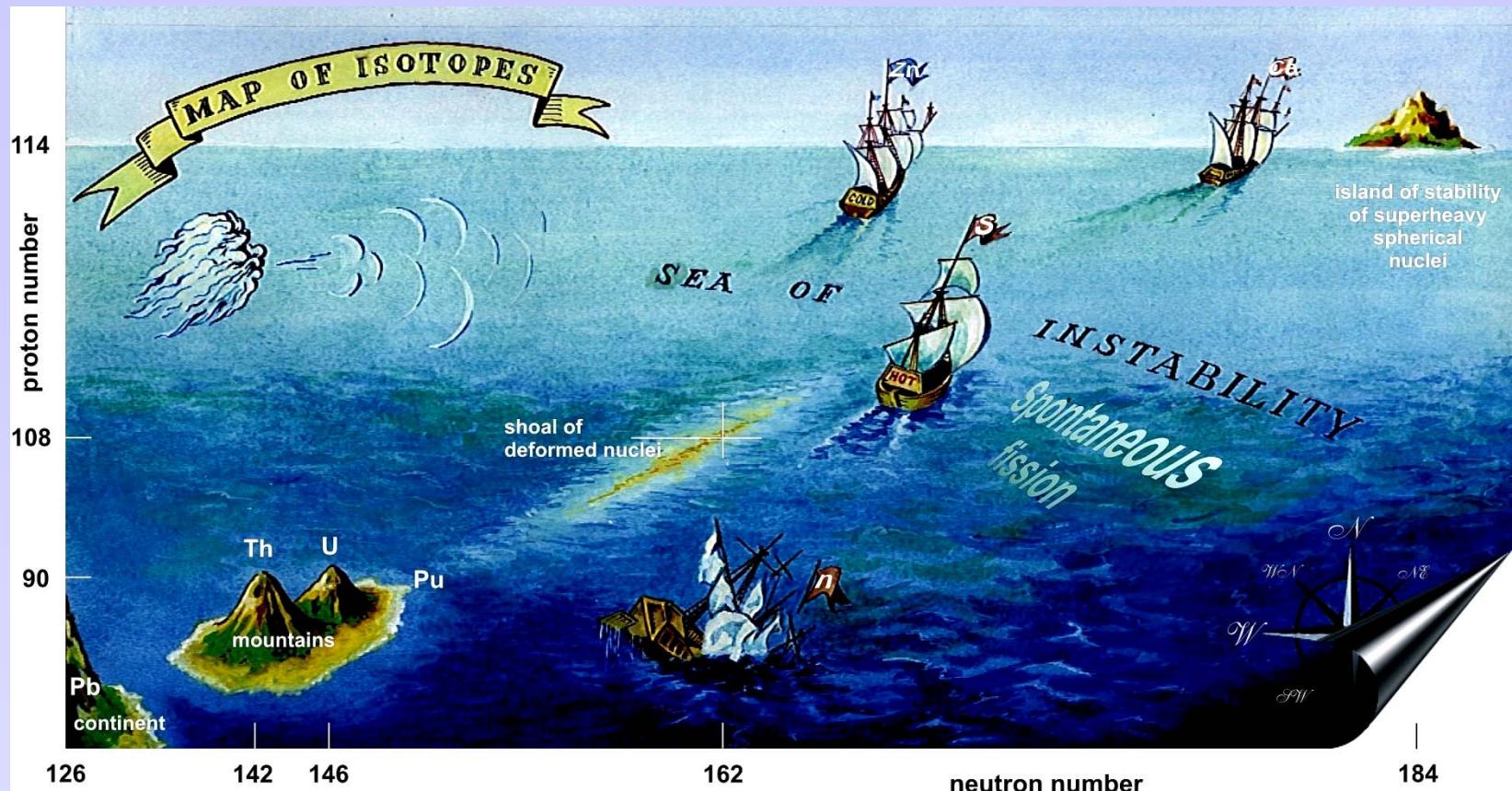
Актиноиды Actinides

Group	90	91	92	93	94	95	96	97	98	99	100	101	102	103
Th		Protactinium	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr
	Zr-90	Zr-91	Zr-92	Zr-93	Zr-94	Zr-95	Zr-96	Zr-97	Zr-98	Zr-99	Zr-100	Zr-101	Zr-102	Zr-103

Н - символ
 1,00794 - атомный номер
 1s - электронная конфигурация
 13,59844 - 1-й потенциал ионизации, эВ
 0,0899 - плотность, кг/м³
 -259,34 - температура плавления, °С
 -252,87 - температура кипения, °С

5-элементы
0-элементы

d-элементы
f-элементы



Thank you for your attention!
Many thanks to the organizers!!